

New Compounds

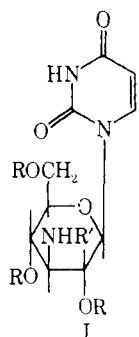
Some Aminoacyl Derivatives of Amino Sugar Nucleosides

HERBERT A. FRIEDMAN

*Department of Pharmacology, University of Pennsylvania,
School of Medicine, Philadelphia, Pennsylvania 19104*

Received August 7, 1970

It was recently reported¹ that amino acids could be condensed with amino sugar nucleosides to give the corresponding aminoacyl derivatives. In the quest for new potential nucleoside antibiotics the L-Lys and the L-Glu- γ -Me ester derivatives of 1-(3-amino-3-deoxy- β -D-glucopyranosyl)uracil were prepared. Their syntheses are given below.



Cbo = C₆H₅CH₂CO

- a, R = Ac; R' = H
- b, R = Ac; R' = Cbo-L-Glu- γ -Me ester
- c, R = OH; R' = Cbo-L-Glu- γ -Me ester
- d, R = OH; R' = L-Glu- γ -Me ester
- e, R = OH, Ac, Ac; R' = di-Cbo-L-Lys
- f, R = OH, Ac, Ac; R' = L-Lys

Experimental Section²

1-(3-N-Carbobenzoxy-L-glutamylamido- γ -methyl Ester 3-Deoxy-2',4',6'-tri-O-acetyl- β -D-glucopyranosyl)uracil (Ib).—1-(3-Amino-3-deoxytri-O-acetyl- β -D-glucopyranosyl)uracil·HCl³ (Ia·HCl) (1.2 g, 2.9 mmoles) was added to 7 ml of dry CHCl₃. Soln took place upon the addition of 0.81 ml (5.8 mmoles) of Et₃N. A second soln was prepd containing N-Cbo-L-Glu- γ -Me ester⁴ (856 mg, 2.9 mmoles), Et₃N (0.41 ml, 2.9 mmoles), and CHCl₃ (8 ml). Both solns were cooled to 0° in an ice bath. Methyl chloroformate (0.22 ml, 2.9 mmoles) was added to the second soln, and the mixture was stirred for 20 min at 0°. The soln contg the nucleoside was added to the mixed anhydride, and the mixture was stirred for an additional 15 min at 0° followed by overnight stirring at room temp. The soln was extd with 0.2 N HCl, followed by a satd soln of NaHCO₃ and finally with H₂O contg a few crystals of NaCl (to discourage emulsion formation). The CHCl₃ layer was dried (MgSO₄) and evapd *in vacuo*. Trituration of the residue with cyclohexane-CHCl₃ (10:1) yielded partially cryst material (450 mg, 23%). It had an indefinite melting point beginning at 105°. Silica gel tlc (CHCl₃-MeOH, 5:1) indicated one spot with a slight trace of impurity. *Anal.* (C₃₀H₃₆N₄O₁₄) C, H, N.

1-(3-N-Carbobenzoxy-L-glutamylamido- γ -methyl Ester 3-Deoxy- β -D-glucopyranosyl)uracil (Ic).—Compound Ib (450 mg)

(1) H. A. Friedman, K. A. Watanabe, and J. J. Fox, *J. Org. Chem.*, **32**, 3775 (1967).

(2) Melting points were taken on a Thomas-Hoover capillary melting point apparatus and are uncorr. Microanalyses were performed by Galbraith Laboratories, Inc., Knoxville, Tenn. Where analyses are indicated only by symbols of the elements, analytical results obtained for those elements were within $\pm 0.4\%$ of the theoretical values.

(3) K. A. Watanabe, and J. J. Fox, *Chem. Pharm. Bull.*, **12**, 975 (1964); K. A. Watanabe, J. Beranek, H. A. Friedman, and J. J. Fox, *J. Org. Chem.*, **30**, 2735 (1965).

(4) Mann Research Laboratories, New York, N.Y.

was added to 15 ml of anhyd MeOH. Following the addition of 7 mg of anhyd LiOH the soln, protected from moisture, was allowed to stir for 26 hr at room temp. LiOH was removed by the addition of a small amt of Dowex 50 (H⁺) resin and stirring the slurry for 1 hr. Silica gel tlc (CHCl₃-MeOH, 5:1) indicated one major component along with a trace of impurity. The compd had an indefinite melting point, yield 320 mg (85% based on Ib). *Anal.* (C₂₄H₃₀N₄O₁₁·H₂O) C, H, N. The H₂O presumably came from the resin.

1-(3-N-L-Glutamylamido- γ -methyl Ester 3-Deoxy- β -D-glucopyranosyl)uracil (Id).—Compd Ic was prepd without the isolation of Ib and was dissolved in 80% aq MeOH. The soln pH was maintained at ca. 1 by adding HCl during hydrogenolysis (10% Pd-C). The product, a tan-colored compd, was isolated by filtration of the catalyst and evapn of the supernatant soln. Recrystn from *n*-PrOH gave a compd in a yield of 280 mg (19% based on Ia). The melting point was of no value as shrinking began at ca. 60° and continued slowly with rising temp. Silica gel tlc (*t*-BuOH-MEK-H₂O-NH₄OH 4:3:2:1) indicated minor impurities similar to those found prior to purification. *Anal.* (C₁₈H₂₂N₄O₉·HCl·C₃H₇OH·H₂O) C, N; H: calcd, 6.43; found, 5.63.

1-(3-N,N'-Dicarbobenzoxy-L-lysylamido-3-deoxydi-O-acetyl- β -D-glucopyranosyl)uracil (Ie).—The procedure followed was similar to that for Ib and Ic except that N,N'-di-Cbo-L-Lys⁵ was used. During the deacetylation reaction with LiOH a ppt developed which was filtered and washed well with Et₂O; yield 18% (based on Ia), mp 110–125°. Silica gel tlc (CHCl₃-MeOH, 5:1) indicated a pure compd. *Anal.* (C₃₆H₄₃N₅O₃·CH₃OH) C, H; N: calcd, 8.92; found 8.43. The results indicate that only one acetate group was removed during the deacetylation reaction, the resultant diacetoxy compd being insol. Prolonged reaction times with increased amts of LiOH did not change the results according to tlc. The position of deacetylation remains uncertain.

1-(3-L-Lysylamido-3-deoxydi-O-acetyl- β -D-glucopyranosyl)uracil Dihydrochloride (If).—The procedure followed was similar to that for Id except that EtOH was used for titration; white solid, yield 67% (based on Ie); mp 196–203°. *Anal.* (C₁₈H₃₁N₃O₂·2HCl·0.5H₂O) C, H; N: calcd, 12.89; found, 12.30.

Acknowledgment.—This study was supported by a grant from the Southeastern Pennsylvania Heart Association.

⁵ National Biochemicals Corp., Cleveland, Ohio

Heterocyclic N-Carboxamides as Anticonvulsants

GIANFRANCO PALA,* ARTURO DONETTI,
ANTONIO MANTEGANI, AND AMEDEO OMODEI SALÉ

*Research Laboratories of Istituto De Angeli,
Milan, Italy*

Received August 21, 1970

In view of the potent anticonvulsant activity of carbamazepine,¹ three new heterocyclic N-carboxamides were synthesized. Anticonvulsant tests have shown that only 1-carbamyl-2,3-diphenylaziridine (**4**) is active, even though its potency was found to be distinctly inferior to that of carbamazepine.

* To whom correspondence should be addressed.

(1) W. Theobald and H. A. Kunz, *Arzneim.-Forsch.*, **13**, 122 (1963).